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9. Use the symbols in Table 11.1 on page 323 to write a skeleton equation for the

following chemical reaction. Hydrochloric acid reacts with zinc to

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produce aqueous zinc(II) chloride and hydrogen gas.

Balancing Chemical Equations (pages 324–328) 10.

## **SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)**

330 Chapter 11 11.2

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barbeque grills through  
the burning of carbon.

Have you ever felt the  
heat and smelled the  
smoke coming from a  
burning charcoal grill?

The heat and smoke  
are the products of a  
combustion reaction.  
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9. Use the symbols in Table 11.1 on page 323 to write a skeleton equation for the following chemical



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reaction. Hydrochloric acid reacts with zinc to produce aqueous zinc(II) chloride and hydrogen gas.

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11.  $2\text{Ni}_3\text{S}_2 \rightarrow 6\text{Ni} + 4\text{S}_2$

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