

## Chapter 11 Assessment The Mole Reviewing Vocabulary

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Section 10.2 Assessment page 332 22. Summarize in terms of particles and mass, one-mole quantities of two different monatomic Each one-mole quantity has 6.02 10<sup>23</sup> particles (atoms), but they will have different masses. 23. State the conversion factors needed to convert between mass and moles of the element Mass-to-mole conversions use the ...

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### Chemistry Chapter 10 The Mole Assessment Answers

308 Chapter 11 The Mole CHAPTER 11 What You'll Learn You will use the mole and molar mass to make conver-sions among moles, mass, and number of representa-tive particles. You will determine the per-cent composition of the components of compounds. You will calculate the empirical and molecular for-mulas for compounds and determine the formulas ...

### Chapter 11: The Mole

Solutions Manual Chemistry: Matter and Change • Chapter 11 211 CHAPTER 11 SOLUTIONS MANUAL Section 11.1 Assessment page 372 5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each ...

### StoichiometryStoichiometry

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Chapter 11 Assessment The Mole Reviewing Vocabulary Section 11.2 Mass and the Mole. Molar mass. The mass in grams of a mole of any pure substance is called its . molar mass. The molar mass of any element is equal to its atomic mass and has the units g/mol. Ex.

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