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8. calcium chloride (75 (OQ- 9. copper
sulfide 10. N D 11. pentaphosphorus
heptanitride) Intermolecular Forces O
Name: 12. List all of the intermolecular
forces that each molecule will
experience and circle the strongest for
each. CO₂

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Unit 6 Study Guide Answer Key

84 Study Guide for An Introduction to Chemistry Section Goals and Introductions As always, the Review Skills section for this chapter is important. A quick review of atomic mass and moles in Section 3.6, the information on ionic and molecular compounds in Section 5.2, and

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monatomic ion charges in Section 5.3 will be useful.

Chapter 6 More on Chemical Compounds

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each statement below, write true or false.

1. Carbon atoms can bond together in straight chains, branched chains, or rings. true

2. Large molecules containing carbon atoms are called micromolecules. false

3.

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matter, and of the elements of the Earth and its atmosphere.

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Unit 5: Periodic Table Study Guide 1. Fill in the valence electrons trend. Pre-AP Chemistry Name: 3 5 -3-7--1 0 How many valence electrons are in each

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Label the charge of the ions each group
forms. 3. Define electronegativity.
Define ionization energy. re-move.-
Label the arrows as Increasing or
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