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Rate of net energy transfer by heat, work, and mass. system (steady) Rate of change in internal, kinetic, potential, etc. energies. out (since $k_e \geq 0$) 12434 124443444 Δ . $\Delta\Delta$. \hat{E} 0. 12 12. 0. Substituting, $Q_{out} = \dot{m} c_p \Delta T = (120 \text{ kg/min})(1.007 \text{ kJ/kg} \cdot ^\circ\text{C})(^\circ\text{C}3) = 363 \text{ kJ/min}$. 120 kg/min AIR. Q · Chapter 1 Basics of Heat Transfer

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1-2. Thermodynamics and Heat Transfer. 1-1C Thermodynamics deals with the amount of heat transfer as a system undergoes a process from one equilibrium state to another.

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Assumptions Heat transfer from the surface of the filament and the bulb of the lamp is uniform .
Analysis (a) The heat transfer surface area and the heat flux on the surface of the filament are $A_{DL} = \pi d_L L = (0.05 \text{ cm})(5 \text{ cm}) = 0.785 \text{ cm}^2$
 $s = \frac{Q}{A} = \frac{1.91 \times 10^4 \text{ W}}{0.785 \text{ cm}^2} = 2.43 \times 10^7 \text{ W/cm}^2$
 $150 \text{ W/s} = A Q q$ & (b) The heat flux on the surface of glass bulb is

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1-13C The rate of heat transfer per unit surface area is called heat flux q . It is related to the rate of heat transfer by $^3 A Q qdA$. 1-14C Energy can be transferred by heat, work, and mass. An energy transfer is heat transfer when its driving force is temperature difference.

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... least in the aspects of heat and/ or mass transfer, analytical/numerical solutions and optimization of the processes The first chapter deals with the description and mass transfer analysis of fixed-bed ... layer on the heat transfer process in high temperature polymerization Chapter is concerned with the description and analysis of heat and mass transfer processes in a fin -and- tube air heater Hydrogen ...

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