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This lab is an excerpt from Heath Chemistry, a textbook of laboratory experiments.

Information regarding

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Laboratory

the lab and detailed procedure are provided on the following pages.

Note: Change Step 2 from g. 0 g of copper (II) chloride to approximately 3.0 g - 4.0 g

Lab 4B (student copy)

To explain the observations obtained by applying Le Chatelier's Principle Apparatus and Materials: Refer to

Online Library Heath Chemistry

Laboratory
Heath Chemistry
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Experiments page 209
Procedure: Refer to
Heath Chemistry
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Experiments page
209-211 Data and
Observations: Part I
Table1: Equilibrium
stresses involving
Thymol Blue Reagent
Added Stress (Ion
Added ...

**Lab 19A - Morgan
Smith and Shaylin**

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**Quartly April
8th, 2016 ...**

If the same thing happened in either experiment, CO₂ was produced by the experiment. In the first experiment, Mg dissolves in HCl to make aqueous MgCl₂ -- there should be no CO₂ liberated. In the...

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Reactions (April . as
one of the four main
types H AZARDS Most
of these solutions are .
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experiment? I don't
know how to make the
table for question # 5.
"From the results of
the temperature at
which a precipitate
dissolved, make up a
table showing the
temperature, the K_{sp}

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and the solubility."

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Lab: 19 Purpose: To use and mix different concentration of solutions of Pb^{2+} and I^- , then to see if ppt is formed. To acquire the K_{sp} value of PbI_2 at room temperature and temperatures above the room temperature. Materials: Please refer to p. 220 of Heath Chemistry Laboratory Experiments.

Procedure: 1. Obtain 12 test tubes and two test tube racks, then

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put 6 test tubes in
each rack.

Answers

**19c.docx - Lab 19
Purpose To use and
mix different ...**

Heath Chemistry Lab
19A. Chemistry 12 -
Lab 19A Flow Chart
(Investigating Chemical
Equilibrium) PART 1:
Put on safety goggles
and lab apron Obtain 2
(Empty, Clean, AND
DRY) 250mL
Erlenmeyer flasks. Add
approx. 100mL of

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DISTILLED water and
1mL of thymol blue
solution to each flask
Record the colour of
this solution in note
book FIRST FLASK: add
a single drop of 0.1M
HCl.

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